Carbon—Carbon Bond Formation Involving Reactions of Alkynes with Group 9 Metals (Ir, Rh, Co): Preparation of Conjugated Olefins[†]

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ABSTRACT

Stable alkynyl complexes of iridium(III) (L_nIr- \equiv -R) that are prepared from the reactions of terminal alkynes readily undergo the intramolecular C–C bond-forming reactions between the alkynyl and adjacent hydrocarbyl ligands to yield conjugated olefins. These reactions are initiated by electrophiles (H⁺, Me⁺) that attack the β carbon of the alkynyl ligand to increase the electrophilicity of the α carbon of the alkynyl ligand. The C–C bond is then formed between the α carbons of the alkynyl and adjacent hydrocarbyl ligands.

Introduction

Carbon–carbon bond formation is probably the most useful transformation in synthetic chemistry. Transition-metal-mediated C–C bond-forming reactions involving alkynes have been extensively investigated on the basis of the fact that they produce a variety of interesting organic compounds, such as conjugated olefins,¹ unsaturated cyclic hydrocarbons,² α,β -unsaturated aldehydes,³ unsaturated cyclic ketones,⁴ enynes,⁵ alkyne polymers,⁶ arenes,⁷ heteroarenes,⁸ and others.⁹

Reactions of alkynes with transition metals and the related reactions of metal complexes containing vinyli-

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dene, allenylidene, allenyl, and propargyl ligands have been summarized recently in excellent reviews that cover the relevant reactions with mostly group 4, 8, and 10 metals. ¹⁰ This account summarizes the C–C bond formation mediated by iridium-alkynyls, mostly based on our recent works.

Alkynes react with transition metals to yield stable complexes in several ways: (i) reactions of internal alkynes mostly yield π -alkyne complexes (1); (ii) those of terminal alkynes frequently yield alkynyl complexes (2); (iii) metallacyclopentadienes (3) are obtained from the reactions of internal and terminal alkynes, depending on the metals and the alkyne substituents; and (iv) alkynes are inserted into the M-L (L: $C=C=CR'_{2}$, 11a C=C-R', 11b H, 12 Cl, 13 SiR'_{3} , 14 NR'_{3} , 1b,8b,15,16c PR'_{3} , 16 and AsR'_{3} 16b,c) bonds to produce alkenyl complexes (4). It is well-known that hydridoalkynyl complexes (2) readily undergo hydride transfer reactions to yield metal vinylidenes (5), which are also obtained directly from the rearrangement of π -coordinated terminal alkynes. 10b Those metal complexes, 1-5, are the key intermediates in the C-C bond-forming reactions mediated by metals that react initially with alkynes.

Reactions of Metal-Alkynyls

Proton-Mediated Coupling between Alkynyl, Alkyl, and Alkenyl Ligands. Metal-alkynyls (2) are in general obtained by oxidative addition of terminal alkynes to metals $(M + H - C \equiv C - R \rightarrow H - M - C \equiv C - R)$ and by ligand (L) substitution reactions with alkynyl groups in the presence

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[†] Dedicated to Professor L. Vaska on the occasion of his 75th birthday.

of terminal alkynes and base, B ($[M-L]^+ + H-C \equiv C-R + B \rightarrow M-C \equiv C-R + BH^+ + L$). These alkynyl ligands are then capable of coupling with other adjacent hydrocarbyl ligands in the presence of a proton, as shown in eq 1. Reactions of 6 containing three different hydrocarbyl groups may provide valuable information on the selectivity of the coupling between the hydrocarbyl ligands. In this example, exclusive coupling of the alkynyl and alkyl ligands occurs in the proton-mediated reactions of 6 to produce the *cis*-bis(alkenyl) complexes 7 (eq 1), whereas

no coupling of the alkynyl and alkenyl groups has been observed. Deuterium labeling experiments (eq 1) clearly suggest that the initial proton attack on the β carbon of the alkynyl group of **6** leads to this C–C coupling. The subsequent C–C coupling between the two *cis*-alkenyl ligands by reductive elimination reactions of **7** to yield 1,3-dienes (**8**) is somewhat surprising, since the alkylalkenyl-alkynyl iridium **6** does not undergo any reductive elimination reaction in the absence of a proton.

The facile reductive elimination of **8** from **7** may be understood by the formation of a relatively stable four coordinated iridium(I) complex IrCl(CO)(PPh₃)₂. It may also be said that the C-C bond formation reaction (reductive elimination of hydrocarbons) does not occur between the hydrocarbyl ligands of complex **6** in the absence of a proton, probably because the Ir-alkynyl bond in **6** is so strong (much stronger than the Ir-vinyl bond in **7**) that any reductive elimination reaction involving the alkynyl ligand may not readily occur.

Competitive protonation at either the α or β carbon of the alkynyl ligands is seen in the reaction of the bis-alkynyl complex **9**. Whereas protonation of the alkynyl group cis to the CH₃ ligand occurs at the β carbon to yield the allenes (**10**), protonation of the alkynyl group trans to the CH₃ ligand also occurs at the α carbon to yield small amounts of the respective alkynes (**11**) (eq 2), ^{1b} which is

clearly confirmed by deuterium labeling experiments. Formation of allenes (10) can be rationalized by the initial protonation at the β carbon of the alkynyl group, followed by methyl-group migration to give an alkenyl species, as shown in eq 1.1b β -Hydrogen elimination from the methyl group of the alkenyl complex (see Scheme 1) generates the allenes.

Protonation at the β carbon of an alkynyl ligand (M—C=CR) apparently increases the electrophilicity of the α carbon (e.g., **12**) and is also well-known to yield metal-vinylidenes (M=C=CHR). Although vinylidene complexes have not been observed in the above reactions, they cannot be excluded and, in fact, are viable intermediates in the reactions of eqs 1 and 2 (eq 3).

$$6, 9 \xrightarrow{H^+} \begin{bmatrix} CH_3 & CH_3 & R^+ \\ H & 12 & 13 \end{bmatrix} \xrightarrow{H_3C} \xrightarrow{R^+} H$$

$$(3)$$

Although complex **7** can be isolated in high yield at the early stage of the reaction of **6** with HCl, further reaction of **7** with HCl at room temperature yields *cis*-CH₃CH=CHR exclusively and the dichloro complex **15** (eq 4). The H/D exchange between the CH₃ group and DCl readily and extensively occurs during the formation of the cis olefins to yield d_5 olefins, *cis*-CD₃CD=CDR (eq 4).

Iridium allenes (H–Ir– η^2 –H₂C=C=CHR in Scheme 1) have been suggested as the intermediates that rapidly undergo H/D exchange with DCl.^{1b} That allenes are not found from the reactions of **6** with HCl and cis olefins are not seen from the reactions of **9** may be understood in terms of relative stability of the metal complexes that are expected to be produced with the formation of allenes (eq 2) and cis olefins (eq 4). Trichlorocarbonyl complex IrCl₃(CO)(PPh₃)₂ and hydridoalkenyl complex [IrHCl(CH=CH–NEt₃)(CO)(PPh₃)₂]⁺ are the expected products if cis olefins and allenes are produced from the reactions of **9** (eq 2) and **6** (eq 4) with HCl, respectively. These two complexes have never been reported thus far and have never been found from our experiments.

Related binuclear alkynyl—alkyl complexes are no exception to the proton-initiated C—C bond formation between the alkynyl and alkyl groups (eq 5). Complex **16**

Scheme 1

 $[Ir] = Ir(-CH=CH-^{\dagger}NEt_3)(CO)(PPh_3)_2$

does not undergo the reductive elimination (C–C coupling between the alkynyl and alkyl groups) in the absence of H^+ . The diene (CH₃CH=CH-p-C₆H₄-CH=CHCH₃, **17**) is obtained from the reaction of **16** with HCl in high yield. ¹⁸ The yield of the diene **17** seems dependent on the trans ligand (-CH=CH- $^+$ PPh₃) to the CH₃ group. ¹⁸ A variety of other products (such as CH₃CH=CH-p-C₆H₄-CH=CH-CH-CH₂Cl, CH₃C=C-p-C₆H₄-CH=CHCH₂Cl, CH₃C=C-p-C₆H₄-CH=CHCH₃, and CH₃C=C-p-C₆H₄-CH=CHCl) are obtained with the trans ligand being CH₃CN. ¹⁸ Identification of the deuterium-containing product (**17**- d_4 , CH₃CD=CD-p-C₆H₄-CD=CDCH₃) unambiguously suggests the same reaction procedure (the initial proton attack on the β carbon of the alkynyl group followed by migration of the CH₃ group to the α carbon) as shown in eq 3.

$$CH_3$$

$$CH_3$$

$$CH_3$$

$$CH_3$$

$$CD$$

$$CH_3$$

$$CD$$

$$CH_3$$

$$CD$$

$$CH_3$$

$$CD$$

$$CH_3$$

$$C$$

An interesting example of C–C bond formation has also been observed in the reaction of HC≡CH with the related binuclear alkyl-iridium(I) complex **18**. ¹⁹ The oxidative addition of HC≡CH to the Ir–CH $_3$ of **18** produces the binuclear H–Ir(III)/Ir(I) complex with μ – η ¹: η ²–C≡CH ligand that is converted into the bridging vinylidene ligand (**20**) at lower temperature. Then, the exchange between the methyl ligand and the hydrogen of the bridging vinylidene ligand in **20** (C–C bond formation between the β carbon of the bridging vinylidene ligand and the methyl

ligand) occurs in the presence of CO to give a hydrido- μ -methylvinylidene complex **21** (eq 6).¹⁹

Somewhat unexpected C-C bond formation has been observed from the reaction of a binuclear alkynyliridacy-clopentadiene, **22**²⁰ (eq 7). Proton attack does not seem

to occur on the β carbon of the alkynyl groups of **22** (no conjugated 1,3,5-trienes, such as CH₂=CH-CH=CH-CH=CH-p-C₆H₄-CH=CH-CH=CH-CH=CH₂, have been observed from the reactions of **22** with HCl).²¹ Instead, the proton seems to attack the α carbons of the alkenyl groups to initiate the C-C bond formation (reductive elimination) to yield the dienyne **23** in high yield, which is supported by the deuterium-containing product dienyne (23- d_2). It seems unprecedented that the proton selectively attacks the α carbon of the alkenyl group of alkynyl-alkenyl complexes.

Alkynes are inserted into Ir—P bonds to yield complexes containing alkenyl ligands Ir—CR=CR'—+PPh₃. ¹⁶ These alkenyl complexes with an adjacent alkyl ligand (**24** and **25**) ^{16a} do not undergo any C—C bond-forming reaction between the alkyl and alkenyl ligands, for which no plausible explanation is provided yet. We have prepared more than two dozen such compounds similar to **24** and **25**, none of which have been found to undergo the C—C coupling reactions between hydrocarbyl ligands under various experimental conditions.

Electrophile (H⁺, Me⁺)-Mediated Coupling between Alkynyl and η^3 - or η^1 -Allyl Ligands. Electrophiles (H⁺, Me⁺) also mediate the coupling between alkynyl and η^3 -allyl groups coordinated to iridium (26) to yield η^3 -pentadienyl complexes 30 (eq 8). The α carbon of the

alkynyl group becomes more electrophilic (as shown in **27**) by the attack of the electrophile on the β carbon of the alkynyl ligand and accepts the nucleophilic terminal carbon of the allyl group to form a new C–C bond. A C–C coupling between the α carbons of the vinylidene and η^1 -

allyl groups may not be excluded, since the coordination of X⁻ to the metal may cause a conversion of the η^3 -allyl complex to the η^1 -allyl one. It is most likely that the η^3 pentadienyl (η^2 -alkene- η^1 -alkenyl) complexes **29** are formed initially immediately after the C-C bond formation step and then undergo the 1,3-hydrogen shift reaction, probably through β -hydrogen elimination, to yield another type of η^3 -pentadienyl complexes **30**, which can be isolated in high yield under controlled reaction conditions.^{1a} The trans, trans-conjugated dienes (31) are obtained quantitatively from the reactions of **30** with aqueous HCl. ^{1a} The terminal carbon of the allyl ligand in 32 also reacts with the newly added alkynyl group in the same manner as shown by eq 8 (26 \rightarrow 31) to yield conjugated 1,3,5-trienes (33) (eq 9).1a It is interesting to notice that the C-C coupling occurs preferably through the terminal carbon of the η^3 -allyl ligand in eqs 8 and 9, whereas a C-C bond is formed through the central carbon of the η^3 -allyl ligand of the related complex $[Cp^*Ir(\eta^3-CH_2CHCH_2)(\eta^2-CR\equiv CR)]^+$ in the reaction with a nucleophile.²²

The reaction of an alkynyl- η^1 -allyl complex **34** with a proton shows an interesting C–C bond formation between the allyl and alkynyl ligands.²³ Proton attack on the γ carbon of the η^1 -allyl ligand induces the coupling between the β carbon of the allyl ligand and the α carbon of the alkynyl ligands to yield enynes **35** (eq 10). It is somewhat surprising not to see the initial attack of the proton on the β carbon of the alkynyl ligand of **34** to initiate the C–C coupling between the alkynyl and α carbon of η^1 -allyl ligand to produce dienes.

Another type of C–C coupling was recently reported between μ - η^1 : η^2 -alkynyl and η^1 -allyl ligands. Warming the binuclear Rh/Ir complexes **36** from -50 to 20 °C causes

 $[Ir] = Ir(Br)(CO)(PPh_3)_2$

an unusual C-C bond formation between the α carbon of the η^1 -allyl ligand and the β carbon of the bridging alkynyl group (eq 11). The π coordination of the alkynyl ligand in **36** seems to make the β carbon electrophilic enough to accept the nucleophilic α carbon of the η^1 -allyl ligand to yield the bridging vinylidene complexes 37.

Unusual copper(III) complexes containing both η^1 -allyl (-CH₂CH=CH₂) and alkynyl (-C≡C-R) ligands were proposed in the catalytic allylation of alkynes as the intermediates that undergo the reductive elimination of 1,4-enynes ($H_2C = CHCH_2 - C = C-R$). The proposed copper(III) intermediates may be so unstable that the typical reductive elimination rapidly occurs between the α carbons of the alkynyl and η^1 -allyl ligands to yield the 1,4enynes.25

It is also interesting to note that the hydrido- η^1 -allylalkynyl rhodium complexes, mer-Rh(H)(-CH2CH=CHR)- $(-C \equiv CR')L_3$ ($L_3 = (PPh_3)_2(CO)$), obtained from the reaction of η^3 -allyl rhodium(I) complexes L₃Rh(η^3 -CH₂CHCHR) with terminal alkynes (H−C≡C−R') selectively undergo the reductive elimination of olefins CH₃CH=CHR but yield no C-C bond formation product resulting from the coupling of the alkynyl and η^1 -allyl ligands.²⁶

An η^5 -pentadienyl complex **38** is obtained from the reaction of complex 26 with protic acid, such as HBF4, in the absence of other bases (such as Cl⁻ and CH₃CN) that readily are coordinated to the metal (eq 12).²⁷ Ring closing (C-C bond formation), however, has not been found between the terminal carbons of the η^5 -pentadienyl group

of 38 in the reaction of complex 26 with the proton. On the other hand, cyclization has been frequently observed from the reactions of η^3 -allyl- π -alkyne complexes of group 9 metals (e.g., **39**) with nucleophiles. 2c,d,22,28,29

$$t$$
-Bu t -Ph t -Ph t -Bu t -Ph t -Bu t -Ph t -Ph

Both cyclic and linear polyenes are produced from the intramolecular coupling reactions between the π -alkyne and η^3 -allyl ligands. 2c,d,22,28,29 Stryker et al. have studied the allyl/alkyne cycloaddition reactions ([3 + 2] and [3 + 2 +2]) in the presence of group 9 metals (Co, Rh, Ir) to prepare five- and seven-membered polyenyl rings, cyclopentenes, and dienes (Scheme 2).2c,d,22 Linear polyenes are also produced in the presence of non-group 9 metals, 29 which is beyond the scope of this Account.

Although an intramolecular C-C coupling occurs between the terminal carbon of the η^3 -allyl and π -alkyne ligands (39 → 43), a new C-C bond is also formed between the central carbon of the η^3 -allyl group and an incoming nucleophile, R' (39 \rightarrow 40).^{22,28b}

The η^3 -1,4-pentadien-1-yl complexes **43** [Cp*Ir(η^3 - $CR = CR - CH_2CH = CH_2)$]⁺, unlike the related η^5 -pentadienyl complex 38 that does not undergo further C-C bond formation reaction,²⁷ further undergo intramolecular (43 → 44) C-C bond forming reactions. It should also be noted that a variety of anomalous rearrangement reactions occur to yield several unexpected products in the reactions of related complexes, for which detailed studies were carried out to elucidate the mechanisms for those rearrangements.2c,d,28a (See the cited references for the detailed mechanisms.) Interestingly, 1,4-dienes (RCH= CR-CH₂CH=CH₂) are obtained in the reactions of **39** with H⁺ in the presence of a ligand CH₃CN.^{28a}

Scheme 2

Other Related Alkyne Transformation Mediated by Group 9 Metals. Vinylidene and allenylidene complexes are obtained from the reactions of alkynes with rhodium and iridium complexes and readily undergo intramolecular C–C coupling reactions when they also have other carbyl ligands adjacent to the vinylidene and allenylidene ligands. For example, rhodium vinylidene—alkyl and—alkenyl complexes 46 yield η^3 -allyl (47) and η^1 -alkenyl (48) complexes through the coupling between the vinylidene and alkyl or alkenyl ligands (eq 13). Iridium

hydrido-alkenyl-vinylidene complexes **49** preferably undergo the C–C coupling between the *cis*-vinylidene and alkenyl ligands to produce 2-alkenyl furan (eq 14), but neither alkyne (HC \equiv CR) nor furan was produced. A variety of unsaturated cyclic compounds are selectively produced by intermolecular C–C bond formation between alkynes and alkenes and by an intramolecular C–C coupling reaction of enynes in the presence of transition metals. $^{33-37}$

Group 9 metal complexes show excellent catalytic activity for both hydroformylation and silylformylation of alkynes,³ as they do for hydroformylation of alkenes.³⁸ Cyclooligomerization of alkynes (cyclization of alkynes to prepare arenes,^{2f,7a,c,39-41} cyclobutadienes/cyclooctatetraenes,^{40c} and fulvenes⁴²), condensation of two alkynes and one nitrile to prepare pyridines,^{8a,43,44} linear oligomerization of alkynes (dimerization of terminal alkynes to prepare enynes,^{1e,5b-h,10b,39a} linear trimerization of alkynes to prepare dienynes^{1c,36a,45}), and polymerization of terminal alkynes⁶ are also very important reactions of alkynes and are known to be successfully mediated by group 9 metal compounds.

Conclusions

This account summarizes some of our recent work involving the electrophile-mediated intramolecular C-C bondforming reactions of newly prepared alkynyl/alkenyl/alkyl/

 η^{3} - and η^{1} -allyl iridium complexes along with other chemistry relevant to our work.

Alkynyl complexes are prepared from reactions of group 9 metals with terminal alkynes. These alkynyl complexes readily undergo C–C bond-forming reactions in the presence of electrophiles that attack the β carbon of the alkynyl ligand to initiate the C–C coupling between the alkynyl ligand and adjacent hydrocarbyl ligands, such as alkyl, alkenyl, and allyl ligands. Further investigation of the selectivity for the C–C coupling between such coordinated hydrocarbyl ligands would provide very useful information that may be utilized in synthetic chemistry, especially in the field of conjugated polyolefins.

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